

PC III

16.4.13

Quantenmechanische Formulierung: Franck-Condon-Prinzip
 Born-Oppenheimer-Näherung: Separation von Elektronen- und Schwingungsbewegung (analog Rotation)

$$E_{\text{ges}} = E_{\text{el}} + E_{\text{vib}} (+ E_{\text{rot}})$$

$$\Psi_{\text{ges}} = \Psi_{\text{el}} \cdot \Psi_{\text{vib}} (\cdot \Psi_{\text{rot}})$$

Übergangsmoment R ($\text{Int} \propto |R|^2$)

$$R = e \int \Psi_{\text{el}}^{I*} \Psi_{\text{vib}}^{I*} \sum_i z_i \vec{r}_i \Psi_{\text{el}}^{\text{II}} \Psi_{\text{vib}}^{\text{II}} d\tau_{\text{el}} d\tau_{\text{vib}}$$

$$\vec{\mu} = e \sum_i z_i \vec{r}_i = \vec{\mu}_{\text{el}} + \vec{\mu}_{\text{kern}}$$

Elektronen- und Kernbeitrag zu $\vec{\mu}$

$$R = \int \Psi_{\text{el}}^{I*} \Psi_{\text{vib}}^{I*} \vec{\mu}_{\text{el}} \Psi_{\text{el}}^{\text{II}} \Psi_{\text{vib}}^{\text{II}} d\tau_{\text{el}} d\tau_{\text{vib}} + \int \Psi_{\text{el}}^{I*} \Psi_{\text{vib}}^{I*} \vec{\mu}_{\text{kern}} \Psi_{\text{el}}^{\text{II}} \Psi_{\text{vib}}^{\text{II}} d\tau_{\text{el}} d\tau_{\text{vib}}$$

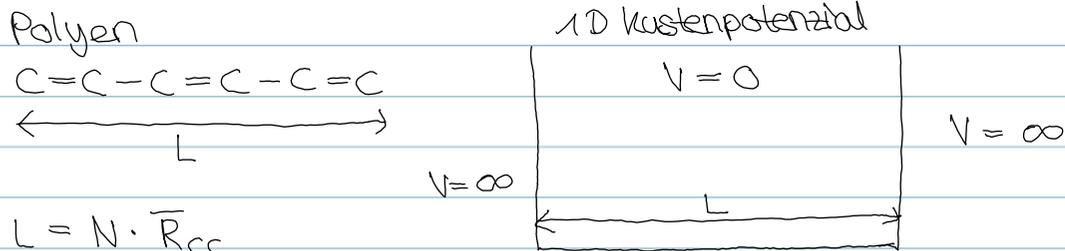
$$\rightarrow \underbrace{\int \Psi_{\text{el}}^{I*} \Psi_{\text{el}}^{\text{II}} d\tau_{\text{el}}}_{=0} \int \Psi_{\text{vib}}^{I*} \vec{\mu}_{\text{kern}} \Psi_{\text{vib}}^{\text{II}} d\tau_{\text{vib}}$$

0 elektronische Wellenfunktionen orthogonal

$$R = \int \Psi_{\text{el}}^{I*} \vec{\mu}_{\text{el}} \Psi_{\text{el}}^{\text{II}} d\tau_{\text{el}} \underbrace{\int \Psi_{\text{vib}}^{I*} \Psi_{\text{vib}}^{\text{II}} d\tau_{\text{vib}}}_{\text{Überlappungsintegral}}$$

Franck-Condon-Faktoren: Quadrat des Überlappungsintegrals bestimmt Intensitätsverteilung

Polyene: Teilchen im Kasten
Beschreibung π -Elektronen

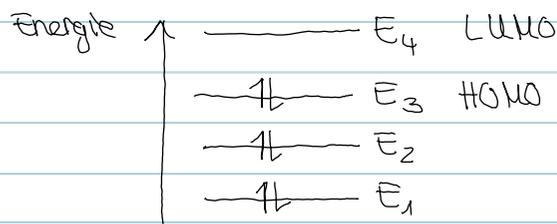


$$L = N \cdot \bar{R}_{CC}$$

$$\bar{R} = 144 \text{ pm}$$

z. B. Hexatrien

1) Bestimme den langwelligsten elektronischen Übergang!



$$L = N \cdot \bar{R}_{CC} = 5 \cdot 144 \text{ pm} = 720 \text{ pm}$$

Hexatrien: 6 πe^-

$$n_{\text{HOMO}} = 3$$

$$n_{\text{LUMO}} = 4$$

$$E_n = \frac{n^2 h^2}{8mL^2}$$

$$\Delta E = E_4 - E_3 = 7 \frac{h^2}{8mL^2} = 8,14 \cdot 10^{-19} \text{ J}$$

$$\Delta E = h \cdot \nu = h \frac{c}{\lambda}$$

$$\lambda = \frac{hc}{\Delta E} = 2,44 \cdot 10^{-7} \text{ m} = 244 \text{ nm (farblos)}$$

2) Bestimme die Normierungskonstante N , wenn

$$\psi(x) = N \sin\left(\frac{n\pi x}{L}\right)$$

$$\sin(\alpha) \sin(\beta) = \frac{1}{2} \cos(\alpha - \beta) - \frac{1}{2} \cos(\alpha + \beta) \quad \text{für } \alpha = \beta$$

$$\alpha \neq \beta$$

$$N^2 \int_0^L \psi^*(x) \psi(x) dx = 1$$

$$N^2 \int_0^L \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{n\pi x}{L}\right) dx = N^2 \frac{1}{2} \int_0^L \cos(0) dx - N^2 \frac{1}{2} \int_0^L \cos\left(\frac{2n\pi x}{L}\right) dx$$

$$= N^2 \frac{1}{2} [x]_0^L - N^2 \frac{1}{2} \frac{L}{2n\pi} \left[\sin\left(\frac{2n\pi x}{L}\right) \right]_0^L$$

$$\frac{N^2}{2} L = 1 \rightarrow N = \sqrt{\frac{2}{L}}$$

$$\psi(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$$

$$\psi_1(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right)$$

$$\psi_2(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{2\pi x}{L}\right)$$

orthogonal?

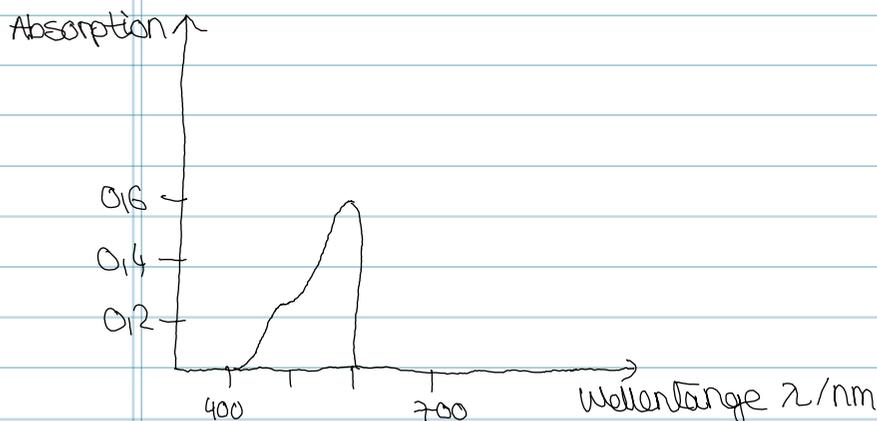
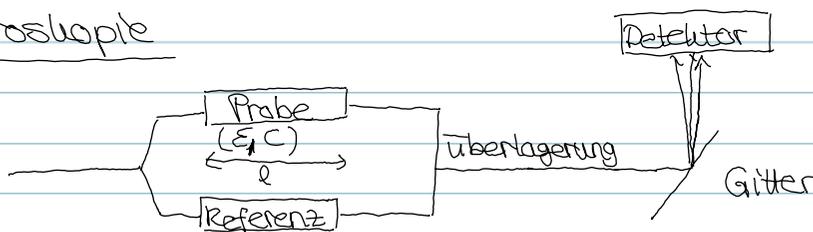
$$\int_0^L \psi_1^*(x) \psi_2(x) dx = \frac{2}{L} \int_0^L \sin\left(\frac{\pi x}{L}\right) \sin\left(\frac{2\pi x}{L}\right) dx = 0$$

UV-VIS - Spektroskopie

Exp.:



Strahlungsquelle



Intensitäten

Transmission $T = \frac{I}{I_0}$

Lambert-Beersches-Gesetz:

$$I = I_0 \cdot 10^{-\epsilon c l}$$

molarer Extinktionskoeff.
Absorptions

Absorption: $A = -\log \frac{I}{I_0} = \epsilon c l$

$$A = \epsilon c l$$